

# Measuring Indoor HCl at the parts-per-trillion (ppt) level with novel Picarro CRDS Gas Concentration Analyzer

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## ABSTRACT

The Picarro SI2108 gas concentration analyzer can provide real-time monitoring of hydrogen chloride (HCl) at a parts-per-trillion (ppt) sensitivity for a variety of applications, including indoor and outdoor air quality, atmospheric science, and hazard assessment.

The Picarro SI2108 operates using Cavity Ring-Down Spectroscopy (CRDS). CRDS utilizes the unique infrared absorption spectrum of gas-phase molecules to quantify isotope ratios and concentrations. Using a small 3-mirrored cavity with a long effective path-length of up to 20 km, measuring the decay rate rather than absolute absorbance, and using a patented wavelength monitor which provides highly accurate measurement and control of the laser's wavelength, allows the instrument to measure HCl at <15 ppt level precision, with a maximum drift over 24 hours of  $\pm 50$  ppt.

Here we present examples of Picarro instruments being used to define sources of HCl indoors. HCl levels inside are generally a few hundred pptv. Using Picarro analyzers, HCl was observed from dishwasher use and electric stovetop use. Reproducible increases in HCl concentrations ( $\sim 100$  pptv) were also observed from bleach applications on a house floor at night, with room lights turned on. The possibility of direct emission or chemical formation of HCl via chlorine atoms is indicated by these observations of HCl derived from household activities.

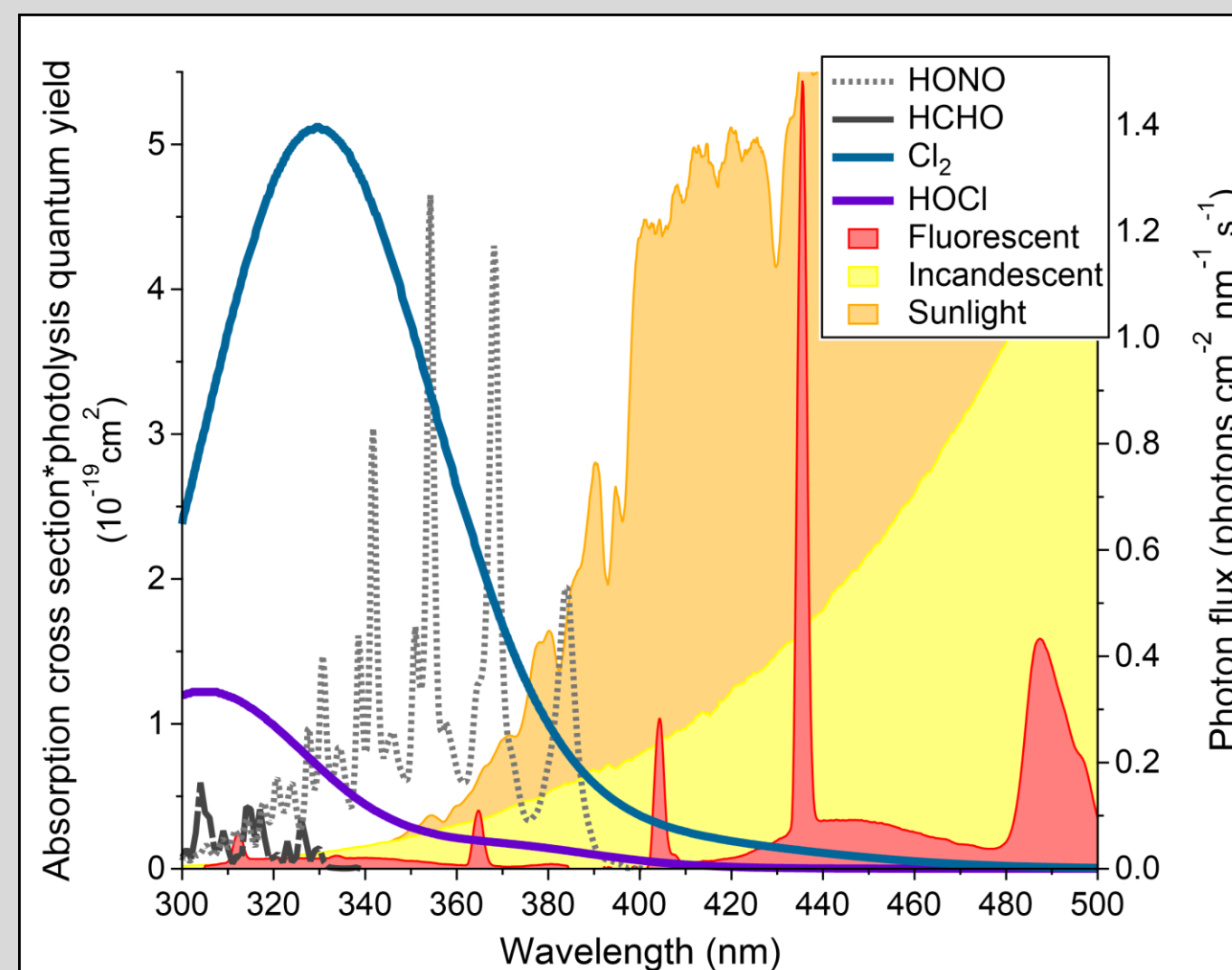
## PICARRO SI2108 CRDS ANALYZER



- $\leq 45$  ppt HCl LDL ( $3\sigma$ , 100 sec)
- $\pm 50$  ppt HCl zero drift (peak-to-peak, 50-minute average)
- 15 ppt HCl precision @  $1\sigma$ , 100 sec
- Coated critical gas pathway components reduces HCl adherence to internal surfaces, which improves response time

## RADICALS INDOORS

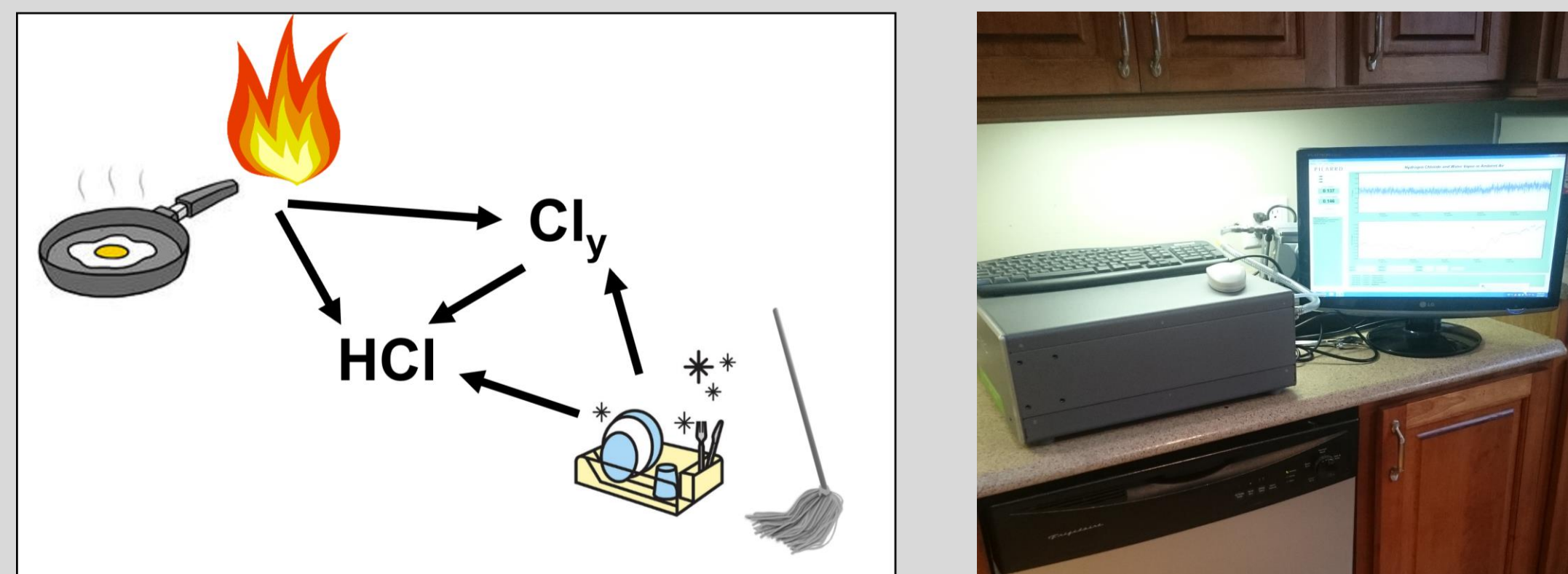
- Lower light available indoors.
  - Photolytic source of OH are less active.
- Indoors, expect greater importance of:
  - $\text{NO}_3$ , which forms in the absence of light.
  - Cl, which has more photolabile precursors.
- Possible that  $\text{Cl}_y$  could have an impact on indoor air quality.



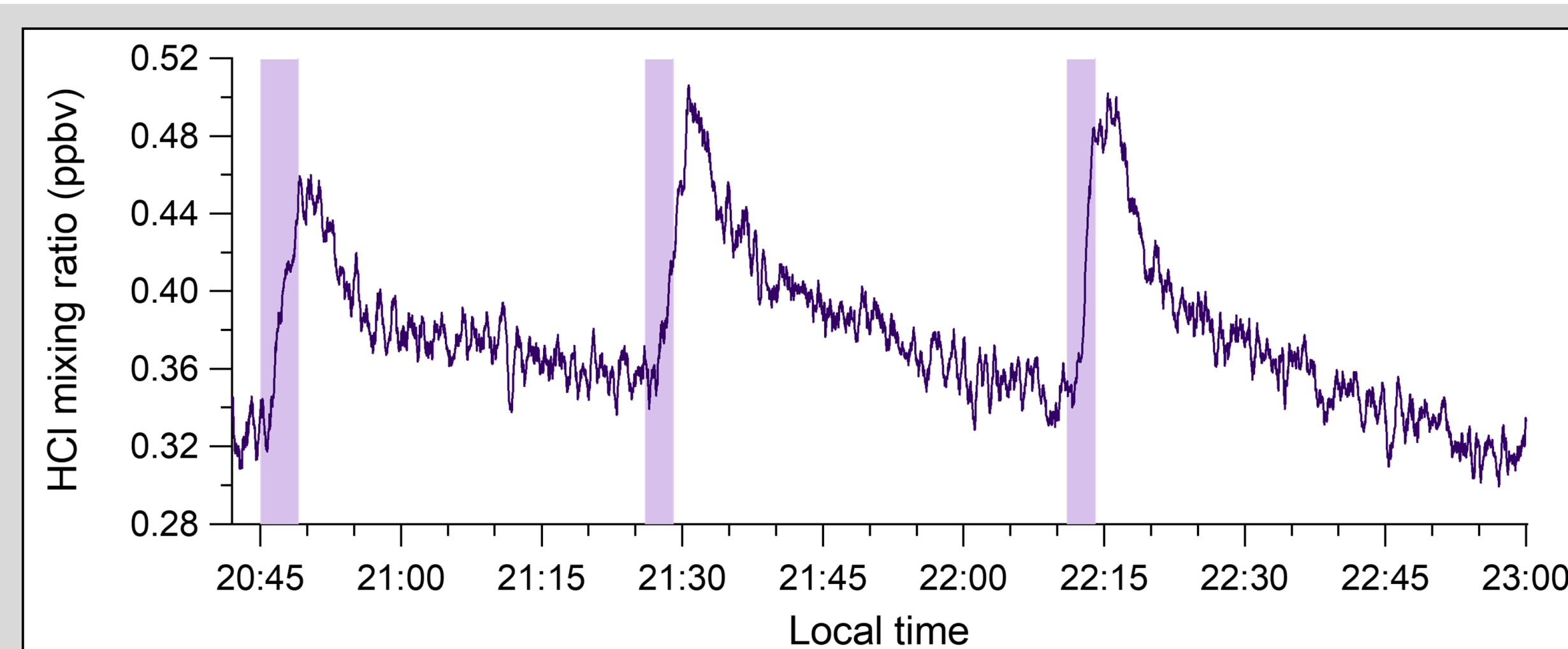
Young et al. 2019 ESPI; Weschler 1992 ES&T

## DIRECT SOURCES OF HCl INDOORS

- Possible direct sources of  $\text{Cl}_y$  include cooking and cleaning activities.
- Measurements of HCl made indoors in an occupied home.



## SOURCES OF HCl INDOORS

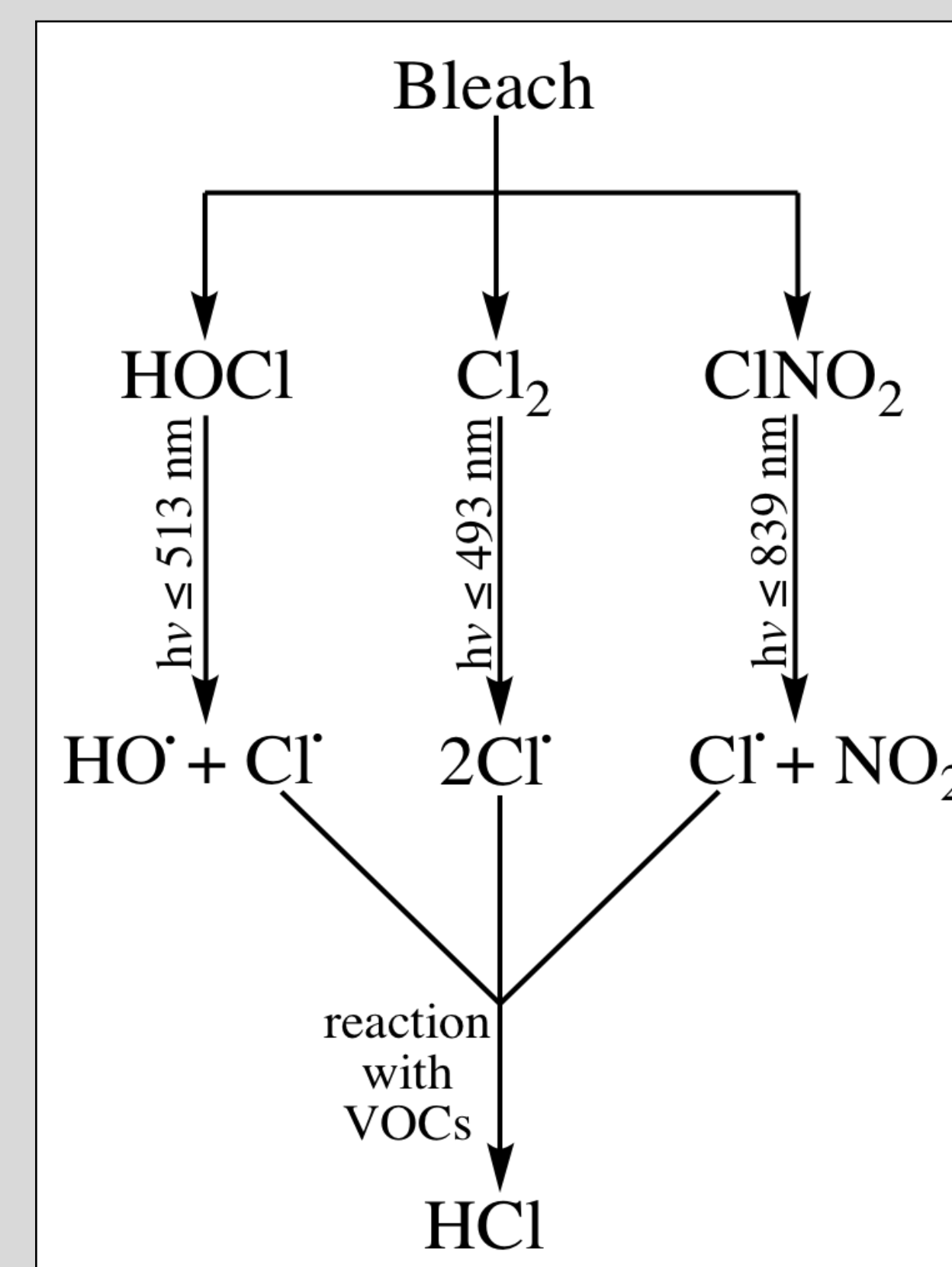


- HCl levels inside generally a few hundred pptv (typically higher than outside).
- HCl observed from dishwasher use (Cl-containing detergent), electric stovetop use.
- Reproducible HCl increases ( $\sim 100$  pptv) observed from repeated bleach applications on a house floor at night, with room lights turned on.
- Calculated HCl production rate  $\sim 1.2 \times 10^7$  molecules  $\text{cm}^{-3} \text{s}^{-1}$ .
- Direct emission or chemical formation possible.

Dawe et al. 2019 Indoor Air

## COULD $\text{Cl}_y$ PHOTOLYZE INDOORS?

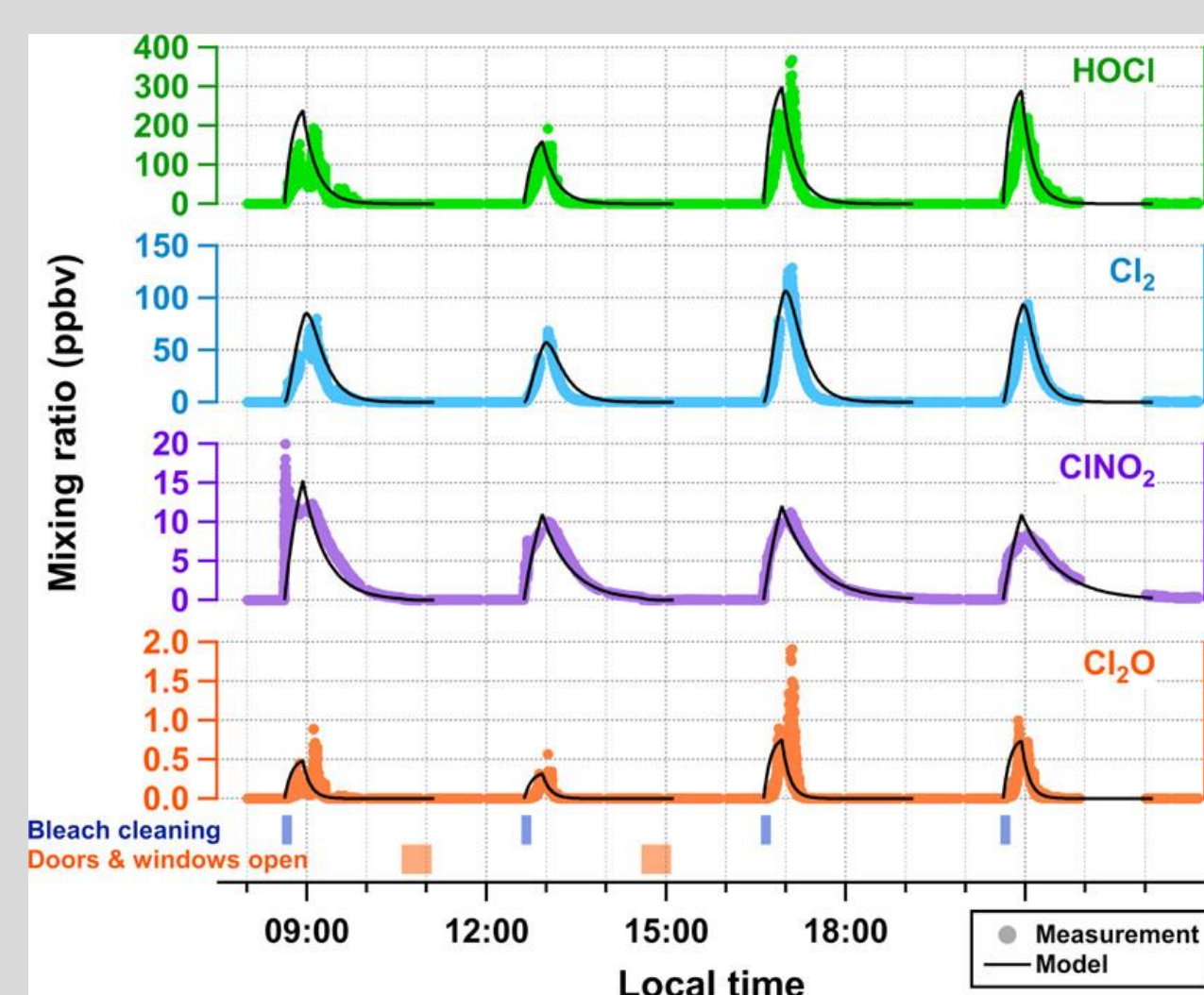
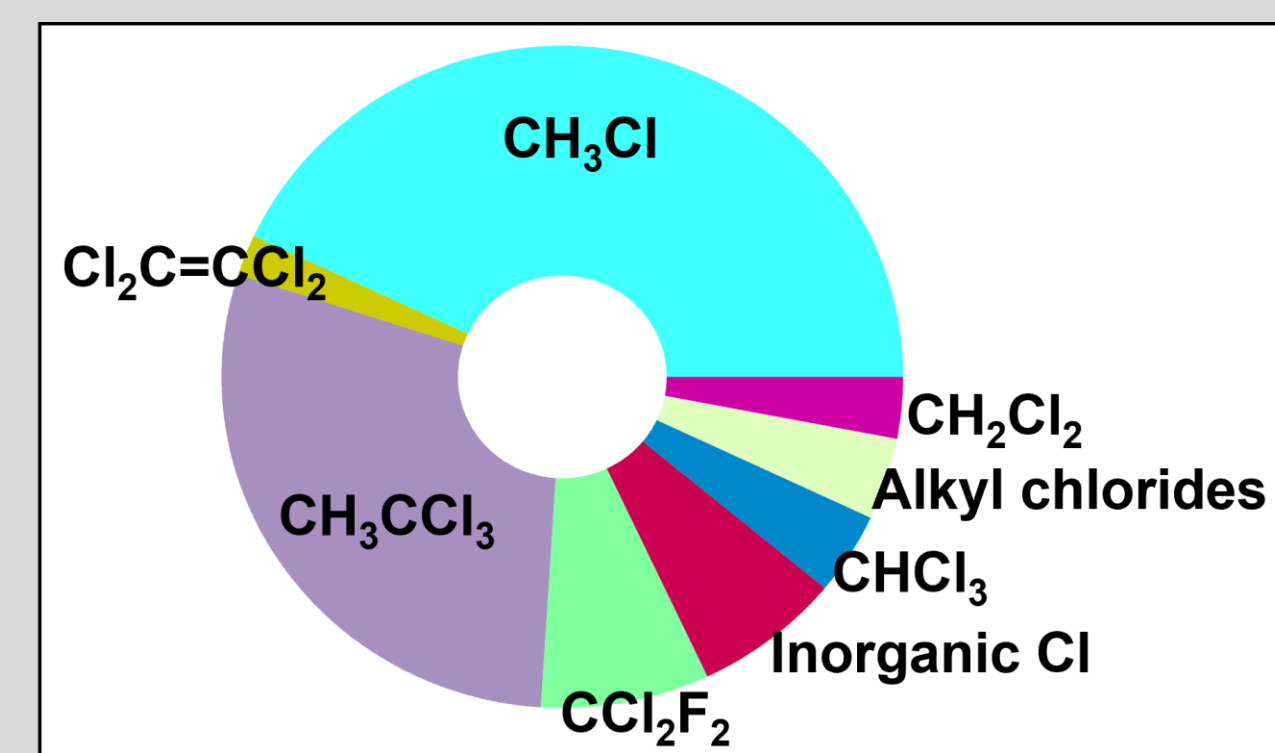
- Sunlit windows could lead to photolysis of some components of  $\text{Cl}_y$ .
- Radiation emitted from LED ( $>400$  nm), halogen ( $>308$  nm), and covered fluorescent tubes ( $>363$  nm) could photolyze reactive Cl.
- Estimated instantaneous Cl production rates:
  - $2 \times 10^2 - 3 \times 10^6$  molecules  $\text{cm}^{-3} \text{s}^{-1}$
  - Similar to predicted OH production rates indoors from HONO photolysis ( $10^5 - 10^7$  molecules  $\text{cm}^{-3} \text{s}^{-1}$ ).
- Chemical formation of HCl is possible.
- We need indoor measurements of all  $\text{Cl}_y$  species simultaneously.



Wong et al. 2017 Indoor Air, Dawe et al. 2019 Indoor Air

## TOTAL GASEOUS Cl ( $\text{TCI}_g$ ) MEASUREMENT

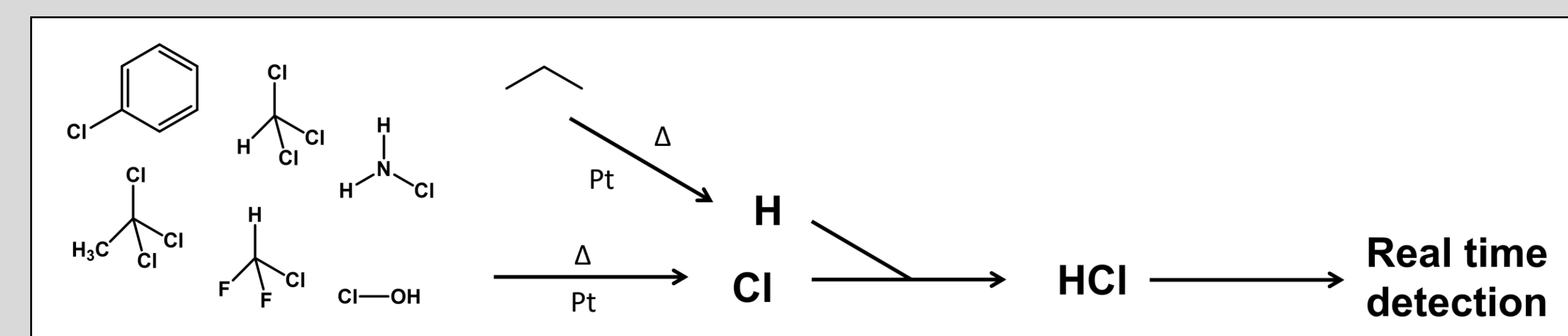
- In recent years, atmospheric chemists have developed methods to measure total C (as  $\text{CO}_2$ ) and N (as  $\text{NO}$ ).
- Coupling different inlets/chemistry can allow simple analyzers to make more complex measurements.
- Analogous to other total elemental measurements, the HCl CRDS can be coupled to a converter to give a total gaseous Cl ( $\text{TCI}_g$ ) measurement.
  - Potential applications to outdoor and indoor environments.



Khalil et al. 1999 JGR Atmos, Mattila et al. 2020 ES&T

## APPROACH TO MEASURING $\text{TCI}_g$

- Approach:
  1. Heated platinum catalyst ( $825^\circ\text{C}$ ) converts  $\text{TCI}_g$  to Cl atoms.
  2. Cl atoms react with H (from  $\text{C}_3\text{H}_8$  provided in excess) to form HCl.

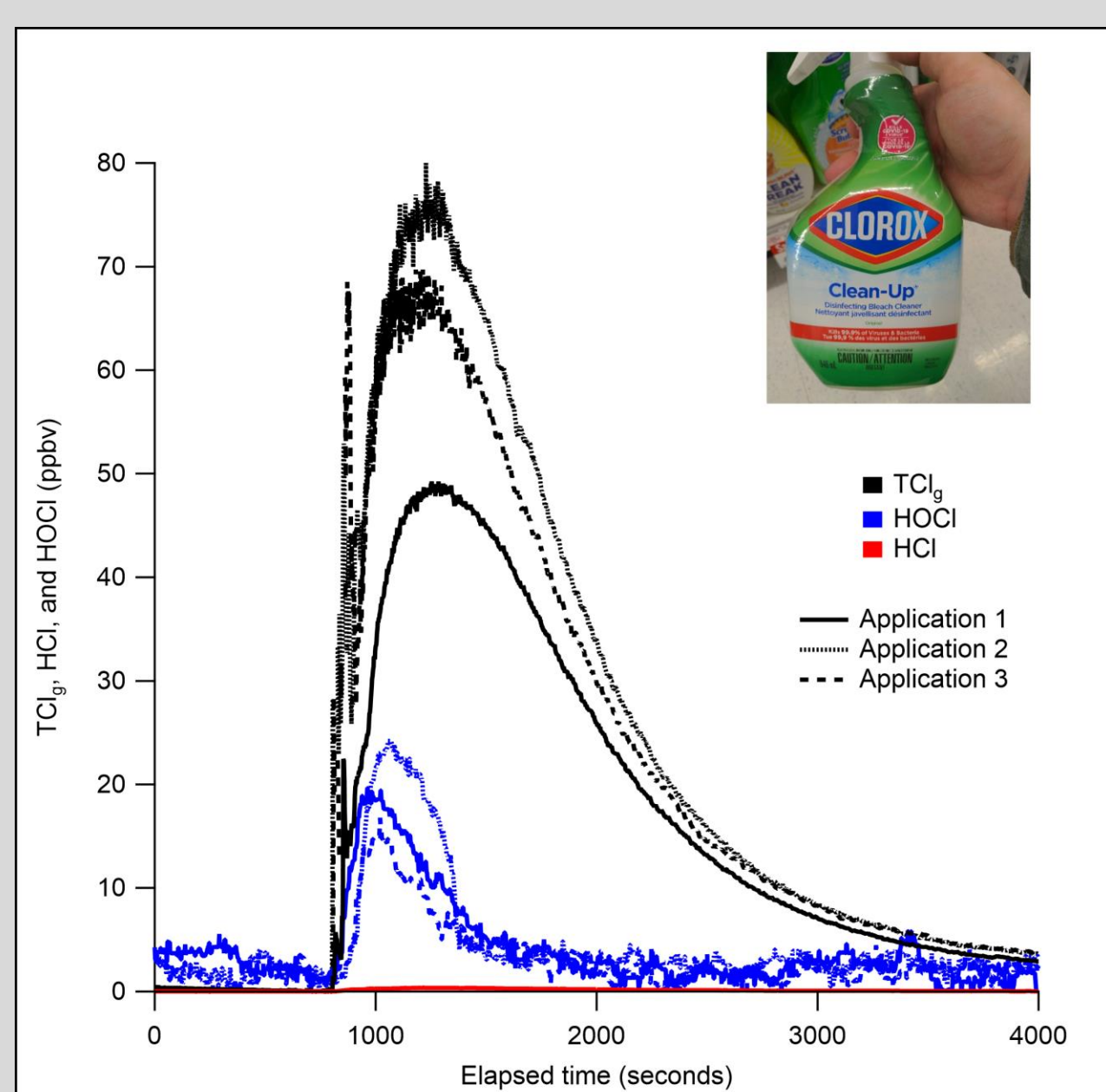


- Ambient air pulled through instrument with residence time of 1.5 seconds.
- Time response of measurement depends on detection limit of HCl method.
- Quantitative response determined using 6 chemicals.

Furlani et al. 2023 AMT

## $\text{TCI}_g$ IN THE INDOOR ATMOSPHERE

- Application of Cl-based spray to floor surface (x 3).
- Co-located HOCl CRDS measurement.
  - $50 - 80$  ppbv  $\text{TCI}_g$ .
  - $15 - 25$  ppbv HOCl.
  - $400$  pptv HCl (separate analysis).
  - Generally consistent with previous measurements.
- Large quantity ( $\sim 80\%$ ) of "missing" chlorine.



Furlani et al. 2022 Atmospheric Measurement Techniques (preprint)